Structure-Based Mutagenesis Reveals Critical Residues in the Transferrin Receptor Participating in the Mechanism of pH-Induced Release of Iron from Human Serum Transferrin

Ashley N. Steere,[†] N. Dennis Chasteen,[‡] Brendan F. Miller,[†] Valerie C. Smith,[§] Ross T. A. MacGillivray,[§] and Anne B. Mason*^{,†}

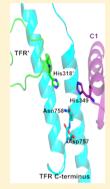
[†]Department of Biochemistry, University of Vermont, College of Medicine, 89 Beaumont Avenue, Burlington, Vermont 05405, United States

[‡]Department of Chemistry, Parsons Hall, University of New Hampshire, Durham, New Hampshire 03824, United States

⁸Department of Biochemistry and Molecular Biology and Centre for Blood Research, University of British Columbia, Vancouver, BC V6T 1Z3, Canada

Supporting Information

ABSTRACT: The recent crystal structure of two monoferric human serum transferrin (Fe_NhTF) molecules bound to the soluble portion of the homodimeric transferrin receptor (sTFR) has provided new details about this binding interaction that dictates the delivery of iron to cells. Specifically, substantial rearrangements in the homodimer interface of the sTFR occur as a result of the binding of the two Fe_NhTF molecules. Mutagenesis of selected residues in the sTFR highlighted in the structure was undertaken to evaluate the effect on function. Elimination of Ca2+ binding in the sTFR by mutating two of four coordinating residues ([E465A,E468A]) results in low production of an unstable and aggregated sTFR. Mutagenesis of two histidines ([H475A,H684A]) at the dimer interface had little effect on the kinetics of release of iron at pH 5.6 from either lobe, reflecting the inaccessibility of this cluster to solvent. Creation of an H318A sTFR mutant allows assignment of a small pH-dependent initial decrease in the magnitude of the fluorescence signal to His318. Removal of the four C-terminal residues of the sTFR, Asp757-Asn758-Glu759-Phe760, eliminates pH-stimulated release of iron from the C-lobe of the Fe,hTF/sTFR Δ 757–760 complex. The inability of this sTFR mutant to bind and stabilize protonated hTF His349 (a pH-inducible



switch) in the C-lobe of hTF accounts for the loss. Collectively, these studies support a model in which a series of pH-induced events involving both TFR residue His318 and hTF residue His349 occurs to promote receptor-stimulated release of iron from the C-lobe of hTF.

he efficient delivery of iron to mammalian cells relies on 📕 the transferrin/transferrin receptor system. Iron (in the form of Fe³⁺) circulates in the blood bound to the \sim 80 kDa bilobal glycoprotein human serum transferrin (hTF). The two lobes of hTF (termed N- and C-lobes) are further divided into two subdomains (N1 and N2, and C1 and C2, respectively). As with most other members of the transferrin superfamily, hTF is capable of binding two Fe³⁺ atoms deep within a cleft formed between the two subdomains of each lobe.^{1,2} The sequestration of Fe³⁺ within the cleft of hTF is critical to preventing its hydrolysis or reduction to Fe²⁺, which promotes the production of harmful radicals via the Fenton series of reactions.^{3,4} Ironcontaining hTFs (diferric and the two monoferric hTF species) bind with nanomolar affinity to the homodimeric transferrin receptor (TFR) located on the surface of many cells.⁵ Following clathrin-dependent endocytosis, a decrease in pH within the early endosome (to pH \sim 5.5–6.0) aids in the release of Fe³⁺ (to an as yet unidentified biological chelator) from the hTF/TFR complex in a receptor-mediated manner.^{6,7} Still within the endosome, the Fe^{3+} must be reduced to Fe^{2+} by a ferrireductase, such as Steap3,8 before being transported out of the endosome by the divalent metal transporter, DMT1.9 Ironfree hTF, termed apohTF, remains tightly bound to the TFR at endosomal pH, facilitating its proper sorting and recycling back to the cell surface.¹⁰ Upon exposure to the more alkaline pH of the blood (\sim 7.4), apohTF is released or displaced by Fe₂hTF from the TFR¹¹ and becomes available for the sequestration of more Fe³⁺.

The TFR plays a critical role throughout the entire process of cellular iron delivery. The release of iron from hTF requires opening of the cleft and is accompanied by large conformational changes within each lobe (opening $\sim 59^{\circ}$ and 50° for the N- and C-lobes, respectively).^{1,2} These significant conformational changes in hTF must be accommodated and compensated for by the TFR within the hTF/TFR complex. As well as being critical to the discrimination between ironcontaining hTF and apohTF at both neutral and endosomal pH, the TFR also significantly affects the release of iron from hTF at endosomal pH.¹² Clearly, the interactions controlling

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Received: January 23, 2012
Revised:
           February 22, 2012
Published: February 22, 2012
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this finely tuned system of cellular iron delivery must be elucidated to understand this process completely.

The crystal structure of the TFR ectodomain [Protein Data Bank (PDB) entry 1CX8]¹³ revealed that the homodimeric receptor is comprised of three distinct domains: a protease-like domain, an apical domain, and a helical domain. Importantly, the TFR crystal structure was used to create a cryo-EM model of the hTF/TFR complex (PDB entry 1SUV).¹⁴ The 7.5 Å resolution cryo-EM model provided the first structural insight into the hTF–TFR interaction. In the model, the N-lobe of hTF is situated between the protease-like domain of the TFR and the cell membrane, while a large portion of the C1 subdomain interacts with the helical domain of the TFR. However, given the relatively low resolution, the model lacked the precision needed to identify specific molecular interactions between hTF and the TFR.

The recently determined 3.22 Å crystal structure of recombinant monoferric N-lobe hTF (Fe_NhTF) bound to the soluble portion of the TFR (sTFR, residues 121-760, PDB entry 3S9L)¹⁵ has provided more detailed information with regard to the binding interactions between hTF and the TFR (Figure 1A). As previously predicted from the crystal structure of the TFR bound to another ligand (the HFE protein),¹⁶ the Fe_NhTF/sTFR structure shows that the structure of the TFR changes significantly as a result of hTF binding. Because Cheng et al.¹⁴ utilized the crystal structure of the unliganded TFR, these structural changes in the TFR as a result of hTF binding were not included in the cryo-EM model. As observed in the HFE/TFR crystal structure,¹⁶ the geometry of a set of four histidine residues (His475 and His684 from each TFR monomer) is altered in the $Fe_NhTF/sTFR$ structure (Figure 1C,D).¹⁵ Specifically, His475 and His475' (which are 7.6 Å apart in the unliganded sTFR structure)¹³ are brought to within 3.6 Å of each other when hTF binds to the TFR.¹⁵ Given the physiologically relevant pK_a of His residues (~6.0), it was suggested that repulsion of this histidine cluster at low pH would promote receptor-mediated iron release and/or release of the HFE protein at endosomal pH.15,16 Additionally, an intersection formed among the apical and protease-like domains of one TFR monomer (TFR), the C-terminus (helical domain) of the other TFR monomer (TFR'), and the C1 subdomain of hTF was identified in the Fe_NhTF/sTFR crystal structure (Figure 1E).¹⁵ A number of interesting structural elements are located within this intersection, including a metal binding site, previously identified in other TFR structures.^{13,16} In the Fe_NhTF/sTFR structure, a Ca²⁺ atom in this metal binding site is coordinated by two residues from the proteaselike domain (Glu465 and Glu468) and three residues from the apical domain of the TFR [Asp307, Thr310, and Phe313 (Figure 1B)]. Another significant feature of this TFR-TFR'-C1 intersection is the large movement of a long loop (residues 275-338) in the apical domain. Specifically, the movement of this loop causes residue His318 to move nearly ~18 Å in the Fe_NhTF/sTFR structure in comparison to the unliganded TFR structure (Figure 1F). Additionally, the location of α -helix 1 in the C-lobe of hTF, on which a number of residues involved in both the binding of hTF to the TFR (Asp356)¹⁷ and pHdependent receptor-mediated release of iron from the C-lobe $(His349)^{15,18}$ are located, is shifted ~5 Å (nearly one full helical turn) in the Fe_NhTF/sTFR structure compared to the cryo-EM model.¹⁵ Although the last two amino acids (Glu759 and Phe760) of the TFR could not be placed in the $Fe_NhTF/sTFR$ structure because of a lack of sufficient density at pH 7.5, it is

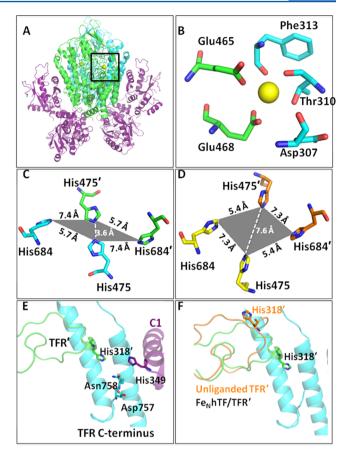


Figure 1. (A) Fe_NhTF/sTFR crystal structure (PDB entry 3S9L).¹⁵ One TFR monomer (TFR) is colored cyan; the other TFR monomer (TFR') is colored green, and the two molecules of Fe_NhTF are colored purple. (B) Ca2+ binding site located between the protease-like (green) and apical (blue) domains of the TFR (A, black box). (C) Histidine cluster formed at the dimer interface. Two His residues, His475 and His684, from one TFR monomer (cyan) converge with the same two His residues, His475' and His684', from the other TFR monomer (green) as a result of hTF binding (in comparison to their position in the unliganded TFR structure, PDB entry 1CX8¹³ as shown in yellow and orange in panel D). (E) TFR-TFR'-Clintersection formed when hTF binds to the sTFR. One TFR monomer is colored cyan, the other TFR monomer (TFR') green, and Fe_{xt}hTF purple. (F) The long loop containing His318 moves ~18 Å upon hTF binding, as shown by overlaying the structure of the unliganded TFR (orange) with the sTFR in the Fe_NhTF/sTFR structure (green).

plausible that α -helix 1 in the C-lobe of hTF could interact with the C-terminus of the TFR monomer at endosomal pH (Figure 1E).

The release of iron from Fe₂hTF can proceed via two pathways: N-lobe first followed by C-lobe $(k_{1N} \rightarrow k_{2C})$, where k_{1N} is the rate constant for the release of the first iron from Fe₂hTF coming from its N-lobe and k_{2C} the rate constant for the release of the second iron coming from its C-lobe) or alternatively release of iron from the C-lobe first followed by the N-lobe $(k_{1C} \rightarrow k_{2N})$. A comprehensive kinetic scheme of removal of iron from diferric hTF (Fe₂hTF) to apohTF was determined by monitoring the increase in intrinsic tryptophan fluorescence as iron is released from recombinant hTF constructs at pH 5.6.^{12,19} In the absence of the sTFR, essentially all iron is released first from the N-lobe followed by the C-lobe.¹² As first suggested by Aisen et al.,²⁰ a switch in the order of iron release is observed in the presence of the TFR, such that iron is preferentially removed from the C-lobe first followed by the N-lobe ($k_{1C} \rightarrow k_{2N}$).¹² However, because this is the case only ~66% of the time, ¹² both pathways must be taken into account when fitting kinetic data from the Fe₂hTF/sTFR complex. Under our defined conditions, the sTFR enhances the release of iron from the C-lobe ~7–11-fold and retards the release of iron from the N-lobe ~6–15-fold.¹² Hence, binding to the TFR not only switches the order of release of iron from hTF but also balances the rates of release of iron from the wolld be account of the two lobes to maximize efficient delivery of iron to cells during the endocytic cycle.

Intrinsic tryptophan fluorescence allows rates of release of iron from each lobe of the hTF/sTFR complexes to be monitored, as well as protein conformational changes. A rapid initial decrease in fluorescence has been a puzzling feature of the kinetic profile of all hTF/sTFR complexes.^{21,22} Because this initial decrease in fluorescence is observed even when the sTFR homodimer alone (without hTF) is exposed to putative endosomal conditions (pH 5.6) and is not observed in any hTF (without the sTFR), it was previously attributed to a pH-sensitive change in the sTFR.²¹ The time points for this initial quench in the tryptophan fluorescence are routinely removed prior to fitting kinetic data for the release of iron from the hTF/ sTFR complexes.¹²

On the basis of the rearrangements within the sTFR observed in the structure of the Fe_NhTF/sTFR complex as discussed above, a number of sTFR mutants have been produced: H318A sTFR, [H475A,H684A] sTFR, [E465A,E468A] sTFR, and a truncated form of the sTFR in which the last four amino acids, Asp757-Asn758-Glu759-Phe760, have been removed (sTFR Δ 757–760). Using our established kinetic scheme for the removal of iron from hTF/sTFR complexes,¹² we have measured the kinetics of release of iron from various recombinant hTF constructs (nonglycosylated Fe₂hTF, Fe_NhTF, and monoferric C-lobe or Fe_ChTF) bound to the sTFR mutants to assess the effect of these mutations on function.

MATERIALS AND METHODS

Materials. Dulbecco's modified Eagle's medium-Ham F-12 nutrient mixture (DMEM-F12) and fetal bovine serum (FBS) were obtained from the GIBCO-BRL Life Technologies Division of Invitrogen. Novex 6% TBE urea mini-gels, TBE running buffer $(5\times)$, TBE-urea sample buffer $(2\times)$, and the iBlot dry blot transfer system were also from Invitrogen. An antibiotic-antimycotic solution (100×) and trypsin were from Mediatech, Inc. The QuikChange site-directed mutagenesis kit was from Stratagene. Pro293A-CDM serum-free medium, L-glutamine, and 4 to 20% acrylamide gels were from Lonza. Methotrexate from Bedford Laboratories was obtained at a hospital pharmacy. All tissue culture dishes, Corning flasks, and expanded surface roller bottles were from local distributors as were Amicon Ultracel 30 kDa molecular mass cutoff membrane microconcentrator devices. Ni-NTA Superflow resin was from Qiagen. Hi-prep 26/60 Sephacryl S-200HR and S-300HR columns were from GE Healthcare. Ethylenediaminetetraacetic acid (EDTA), Ponceau stain, and Orange G were from Fisher. Nitrilotriacetic acid (NTA) and ferrous ammonium sulfate were from Sigma. The chemiluminescence detection kit was from Thermo Scientific.

Expression and Purification of sTFR Mutants. Recombinant N-terminally His-tagged glycosylated sTFR and

various mutants were produced as previously described.²³ Briefly, mutations were introduced into the pNUT vector using the QuikChange site-directed mutagenesis kit. The mutagenic primers are listed in Table 2 of the Supporting Information. Following transfection and selection with methotrexate, adherent baby hamster kidney (BHK) cells containing the mutant N-His-tagged sTFR pNUT vector were transferred into expanded surface roller bottles.²³ The culture medium (~200 mL/roller bottle) was collected every 2-4 days. The first two or three batches that contained DMEM-F12 with an antibiotic-antimycotic solution and 10% fetal bovine serum were collected and discarded. Subsequent batches (generally four to six) containing Pro293A-CDM serum-free medium with L-glutamine and 1 mM butyric acid were collected, pooled, and held at 4 °C until they were purified. The amount of each sTFR mutant produced was determined by a solid-phase competitive immunoassay as previously described.²³

Purification of the sTFR mutants followed the same protocol developed for the sTFR.²³ Briefly, purification entailed concentration followed by the addition of 5× buffer to yield a final concentration of 1× Qiagen start buffer [50 mM Tris (pH 7.5) containing 300 mM NaCl, 20 mM imidazole, 10% glycerol, and 0.05% NaN₃] before the samples were passed over a Ni-NTA column (1 cm × 10 cm) at a flow rate of 2 mL/min. Each sTFR sample was displaced from the column by the addition of 250 mM imidazole to the start buffer. Peak fractions were pooled, reduced using microconcentrators to less than 2 mL, filtered, and loaded onto a Sephacryl S300HR 26/60 column equilibrated and run in 100 mM NH₄HCO₃ at a flow rate of 1.5 mL/min ($V_0 \sim 93$ mL). Fractions containing the sTFR (or mutant) were pooled and stored at 4 °C in 100 mM NH₄HCO₃.

All recombinant nonglycosylated hTFs [diferric hTF (Fe_2hTF), Fe_NhTF (monoferric N-lobe hTF in which mutation of iron binding ligands, Y426F and Y517F, prevents iron binding in the C-lobe), and Fe_ChTF (monoferric C-lobe hTF in which mutation of iron binding ligands, Y95F and Y188F, prevents iron binding in the N-lobe)] were produced as previously described.²⁴

Immunoblotting. Nonreduced sTFR samples were separated by sodium dodecyl sulfate–polyacrylamide gel electrophoresis (SDS–PAGE) on 4 to 20% acrylamide gels preelectrophoresed (20 min at 100 V) with Orange G buffer (0.25% Orange G and 30% glycerol). Samples were then loaded and electrophoresed on the gel (1.75 h at 120 V). Proteins were transferred to nitrocellulose using the iBlot dry blot transfer system. Transfer of proteins to the membrane was confirmed by staining with Ponceau stain. The immunoblot was analyzed using the mouse IgG1 monoclonal antibody to the TFR, A4A6 [1 μ g/mL, a generous gift from the laboratory of J. Cook at the University of Kansas Medical Center (Kansas City, KS)]. Bound antibody was detected using horseradish peroxidase-conjugated goat anti-mouse IgG and a chemiluminescence detection kit.

Formation and Purification of the hTF/sTFR Complex. The hTF/sTFR complexes were prepared by adding a small molar excess (~20%) of hTF (Fe₂hTF or Fe_ChTF) to 1.5 mg of each mutant sTFR. Following equilibration at room temperature for ~5 min, hTF/mutant sTFR complexes were purified by being passed over a Sephacryl S300HR gel filtration column in 100 mM NH₄HCO₃ to remove excess hTF. Fractions containing the complex were concentrated to 15 mg/mL with respect to hTF. Kinetics of the Release of Iron from hTF/sTFR Complexes at pH 5.6. Release of iron from the hTF/mutant sTFR complexes was monitored at 25 °C using an Applied Photophysics SX.20MV stopped-flow spectrofluorimeter as previously described.^{12,18} One syringe contained the hTF/ sTFR complex (375 nM) in 300 mM KCl, and the other syringe contained MES buffer (200 mM, pH 5.6), KCl (300 mM), and EDTA (8 mM). Rate constants were determined by fitting the increase in fluorescence intensity versus time using Origin (version 7.5) to standard models as described in detail previously.^{12,18} When determining rate constants for iron release, we removed the initial quench in the tryptophan fluorescence, attributed to a pH-inducible conformational change in the sTFR, prior to fitting. All data were corrected to zero fluorescence intensity before fitting.

Analysis of the initial quench in tryptophan fluorescence required the derivation of a new model that is similar to the previously described $A \rightarrow B$ model but also includes an initial decay term (Figure 2 of the Supporting Information). The equation used to fit the initial decay data, as well as the complete derivation and program code for Origin, is provided in the Supporting Information. Again, all data were corrected to zero at the fluorescence minimum before fitting. Because the increase in fluorescence following the minimum affects the fit, an equal number of data points on each side of the fluorescence minimum were included in the fitting process. The half-life $(t_{1/2})$ was calculated with the equation $t_{1/2} = \ln(2)/k_2$.

Urea Gel Analysis of hTF/Mutant sTFR Complexes. The iron status of hTF bound to the sTFR mutants was examined by urea gel electrophoresis using Novex 6% TBEurea mini-gels in 90 mM Tris-borate (pH 8.4) containing 16 mM EDTA as previously described.^{12,18} Iron-containing complexes were mixed 1:1 with 2× TBE-urea gel sample buffer (final concentration of 0.5 μ g/ μ L). To determine the extent of removal of iron from the various hTF/mutant sTFR complexes, an aliquot of each was added to iron removal buffer [100 mM MES buffer (pH 5.6) containing 300 mM KCl and 4 mM EDTA] and incubated at room temperature for 5 min. The iron removal process was halted by addition of 2× TBE-urea gel sample buffer. Samples (3.0 μ g) were loaded, and the gel was electrophoresed for 2.25 h at 125 V. Protein bands were visualized by staining with Coomassie blue.

RESULTS

Production of Recombinant sTFR. As reported previously, the expression of the glycosylated sTFR in our BHK system generally produces between 30 and 40 mg of protein per liter of tissue culture medium (Table 1 of the Supporting Information).²³ Production of the H318A sTFR and sTFR Δ 757–760 mutants was comparable to production of the wild-type sTFR, while an ~50% decrease was observed in the production of the [H475A,H684A] sTFR mutant. However, mutation of the two glutamate residues in the protease-like domain of the sTFR ([E465A,E468A] sTFR mutant) significantly decreased the yield of receptor to <1.0 mg/L.

In the final step of purification, homodimeric wild-type sTFR typically elutes from an S300HR size exclusion column as a broad peak centered at 165 mL.²³ However, the [E465A,E468A] sTFR mutant (as indicated by monitoring the A_{280}) eluted from the column immediately following the void volume (V_0) of 93 mL (data not shown), indicating that it exists largely in an oligomeric state. While the wild-type sTFR migrates as a monomer (Figure 2, lane 1) even under

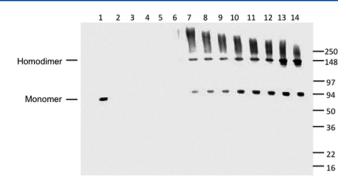


Figure 2. Ca^{2+} coordination by Glu465 and Glu468 is critical to the structural integrity of the sTFR. Purified sTFR (100 ng, lane 1) served as a standard. Following purification, aliquots from tissue culture batches 4, 5, and 6 (lanes 2–4, respectively) as well as the pool of batches 4–6 (lane 5). Fractions 3–11 (corresponding to elution volumes between 102 and 126 mL) from the S300HR column of the [E465A,E468A] sTFR mutant were analyzed by SDS–PAGE followed by immunoblotting (lanes 6–14, respectively). Note that homodimeric wild-type sTFR typically elutes as a broad peak centered at 165 mL.

nonreducing SDS–PAGE conditions, immunoblot analysis confirmed the presence of homodimers as well as a range of higher-order oligomers in the S300HR fractions of the [E465A,E468A] sTFR mutant (Figure 2, lanes 7–14). Unfortunately, the poor yield and oligomeric state of the [E465A,E468A] sTFR mutant precluded further experiments with this mutant.

pH-Inducible Changes in the sTFR. As described in the introductory section, for all sTFR-containing samples, an initial rapid decrease in tryptophan fluorescence precedes the increase associated with the release of iron from Fe₂hTF. Interestingly, the half-life of this pH-mediated rapid decrease in fluorescence of sTFR depends on whether Fe₂hTF is bound [Fe₂hTF/sTFR complex vs sTFR alone (Figure 3)]. The H318A sTFR and

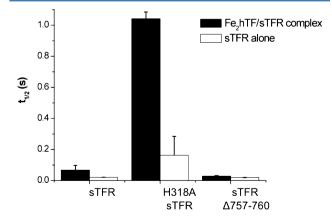


Figure 3. Half-lives $(t_{1/2})$ of a pH-inducible conformational change in the sTFR altered by the H318A sTFR mutation and the sTFR Δ 757–760 truncation. Average half-lives ± 95% confidence intervals are shown for the sTFR, H318A sTFR, and sTFR Δ 757–760 alone (white) and in complex with Fe₂hTF (black).

sTFR Δ 757–760 mutations significantly affect the duration of this pH-sensitive change in the sTFR. Specifically, this pH-sensitive decrease in fluorescence is considerably lengthened in the presence of the H318A sTFR mutation, the half-life being ~8–15-fold longer, depending on whether Fe₂hTF is bound (Figure 3). Conversely, the half-life of the pH-sensitive

Table 1. Kine	tics of the Re	ease of Iron from	H318A sTFR	Complexes"
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complex	$k_{1C} (\min^{-1})$	$k_{2N} (\min^{-1})$	$k_{1\rm N}~({\rm min}^{-1})$	$k_{\rm 2C}~({\rm min}^{-1})$
Fe ₂ hTF/sTFR ^b	5.5 ± 0.9	1.4 ± 0.2	2.8	7.2
Fe ₂ hTF/H318A sTFR	1.8 ± 0.7	1.5 ± 0.2	1.7 ± 0.1	5.9 ± 0.9
complex	$k_1 \;(\min^{-1})$ f	for conformational change	k_{2C} (min	⁻¹) for Fe ³⁺ release
Fe _C hTF/sTFR ^b		20.6 ± 1.2		7.2 ± 0.4
Fe _C hTF/H318A sTFR		-		5.5 ± 0.6
complex	$k_1 \;(\min^{-1})$ f	for conformational change	k_{2N} (min	⁻¹) for Fe ³⁺ release
Fe _N hTF/sTFR ^b		22.0 ± 0.7		1.7 ± 0.6
Fe _N hTF/H318A sTFR		17.4 ± 2.4		1.1 ± 0.1

"Averages and 95% confidence intervals for kinetic runs performed on three to five different days. Each day, three kinetic traces were averaged before fitting." From ref 12.

decrease in the fluorescence is slightly shorter in the presence of the sTFR Δ 757–760 truncation and is unaffected by the presence of Fe₂hTF (Figure 3). No effect on the pH-sensitive fluorescence decrease was observed in the [H475A,H684A] sTFR mutant, with or without Fe₂hTF, in comparison to that of wild-type sTFR (data not shown).

Kinetics of the Release of Iron from hTF/H318A sTFR Complexes. Rate constants were determined by fitting the increase in fluorescence intensity versus time. This increase in intrinsic tryptophan fluorescence has been previously attributed to Trp residues in hTF and not the sTFR.²¹ Because no Trp or Tyr residues in hTF were mutated, no differences in the amplitude of the fluorescence signal in any of the hTF/mutant sTFR complexes in comparison to those of the hTF/wt sTFR complex were observed. TFR residue His318, part of a long loop in the apical domain, flips into the TFR-TFR'-C1 intersection upon hTF binding, moving nearly 18 Å relative to its position in the unliganded TFR structure (Figure 1F). Kinetic rate constants for conformational changes and the release of iron from various hTF/H318A sTFR complexes are presented in Table 1. In the fit of the Fe₂hTF/H318A sTFR kinetic data, rate constants for both pathways, $k_{1N} \rightarrow k_{2C}$ and $k_{1C} \rightarrow k_{2N}$, were allowed to vary (Table 1). The rate constants k_{1C} , k_{1N} , and k_{2C} are smaller in the Fe₂hTF/H318A sTFR complex by 67, 39, and 18%, respectively, relative to the control, while rate constant $k_{\rm 2N}$ is unaffected. The $k_{\rm 1C}/(k_{\rm 1C}$ + $k_{
m 1N}$) ratio corresponds to the fraction following the $k_{
m 1C}
ightarrow k_{
m 2N}$ pathway. For the wild-type Fe₂hTF/sTFR complex, 66% of the iron is removed by the $k_{1C} \rightarrow k_{2N}$ pathway and 34% by the $k_{\rm 1N} \rightarrow k_{\rm 2C}$ pathway. In contrast, the rate constants for the Fe2hTF/H318A sTFR complex indicate that neither pathway is favored over the other; i.e., 51% of the iron is released via the $k_{1C} \rightarrow k_{2N}$ pathway and 49% via the $k_{1N} \rightarrow k_{2C}$ pathway.

Normally, a rapid conformational change precedes the release of iron from the two monoferric complexes, $Fe_NhTF/$ sTFR and $Fe_ChTF/sTFR$, exhibiting conformational rate constants (k) of 22.0 and 20.6 min⁻¹, respectively. In the $Fe_NhTF/H318A$ sTFR complex, a small decrease in the rate of the conformational change (from 22.0 to 17.4 min⁻¹) is observed, whereas the conformational change is completely absent in the $Fe_ChTF/H318A$ sTFR complex. The rate constant for the release of iron from the $Fe_NhTF/H318A$ sTFR complex. The rate constant for the release of iron from the $Fe_NhTF/H318A$ sTFR complex (k_{2N}) is decreased somewhat [from 1.7 to 1.1 min⁻¹ (Table 1)]. Although the preceding conformational change is absent in the $Fe_ChTF/H318A$ sTFR complex, the release of iron from this mutant is relatively unaffected; i.e., this k_{2C} value (5.5 ± 0.6 min⁻¹) is similar to the k_{2C} value obtained from fitting the $Fe_2hTF/H318A$ sTFR complex

 $(5.9 \pm 0.9 \text{ min}^{-1})$ and only 23% lower than the k_{2C} value of 7.2 min⁻¹ for the Fe_ChTF/sTFR complex (Table 1).

Kinetics of the Release of Iron from hTF/ [H475A,H684A] sTFR Complexes. Rate constants for iron release and conformational changes from the various hTF/ [H475A,H684A] sTFR complexes are reported in Table 2.

Table 2. Kinetics of the Release of Iron from [H475A,H684A] sTFR Complexes^a

complex	$k_{1\mathrm{C}}~(\mathrm{min}^{-1})$	$k_{2N} (\min^{-1})$
Fe ₂ hTF/sTFR ^b	5.5 ± 0.9	1.4 ± 0.2
Fe ₂ hTF/[H475A,H684A] sTFR	3.1 ± 0.1	1.2 ± 0.1
complex	$k_1 \; ({ m min}^{-1}) \; { m for} \; { m conformational} \; { m change}$	$k_{2C} \ (min^{-1}) \ for Fe^{3+} \ release$
Fe _C hTF/sTFR ^b	20.6 ± 1.2	7.2 ± 0.4
Fe _C hTF/[H475A,H684A] sTFR	16.7 ± 2.2	7.5 ± 0.9
complex	$k_1 \ ({ m min}^{-1}) \ { m for} \ { m conformational} \ { m change}$	$k_{ m 2N} \ ({ m min}^{-1}) \ { m for} \ { m Fe}^{3+} \ { m release}$
Fe _N hTF/sTFR ^b	22.0 ± 0.7	1.7 ± 0.6
Fe _N hTF/[H475A,H684A] sTFR	15.3 ± 0.3	1.7 ± 0.1

^{*a*}Averages and 95% confidence intervals for kinetic runs performed on three to five different days. Each day, three kinetic traces were averaged before fitting. ^{*b*}From ref 12.

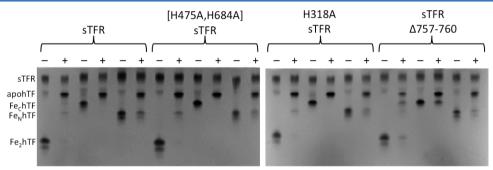
Relatively small differences are observed between the hTF/ [H475A,H684A] mutant sTFR complexes and the respective control complexes, except k_{1C} for the Fe₂hTF complex is reduced by ~45%.

Kinetics of the Release of Iron from hTF/sTFR Δ 757– 760 Complexes. The C-terminal residues of one TFR monomer are positioned between and make contact with both the C1 subdomain of hTF and the other TFR monomer [designated TFR' (Figure 1E)] in the TFR-TFR'-C1 intersection. Kinetic rate constants for conformational changes and the release of iron from various hTF/sTFR Δ 757-760 complexes are listed in Table 3. We were unable to fit the kinetic data for the Fe₂hTF/sTFR Δ 757–760 complex using the rate constants and two-pathway model used for the Fe2hTF/sTFR control complex. Instead, the release of iron from the Fe2hTF/ sTFR $\Delta 757-760$ complex was preceded by a very rapid conformational change (k_1) , followed by the release of iron from the N-lobe $(k_2 = k_{1N})$ and the very slow release of iron from the C-lobe of hTF ($k_3 = k_{2C}$). The specific assignment of these rate constants is established by urea gel analysis of the Fe2hTF/ sTFR Δ 757–760 complex (Figure 4), which indicates that, because of the small rate constant ($k_{2C} = 1.0 \text{ min}^{-1}$), a population of Fe_ChTF remains after the Fe₂hTF/sTFR Δ 757–760 complex is

Table 3. Kinetics of the Release of Iron f	rom sTFR Δ 757–760 Complexes"
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complex	$k_{1\mathrm{C}} \; (\mathrm{min}^{-1})$		$k_{2N} (\min^{-1})$
Fe ₂ hTF/sTFR ^b	5.5 ± 0.9		1.4 ± 0.2
complex	$k_1 \ (\min^{-1})$ for conformational change	$k_2 = k_{1N} (\min^{-1})$	$k_3 = k_{2C} \; (\min^{-1})$
H349A Fe ₂ hTF/sTFR ^c	23.7 ± 4.6	6.7 ± 0.3	0.61 ± 0.02
$Fe_2hTF/sTFR \Delta 757-760$	89.4 ± 2.0	5.2 ± 0.2	1.0 ± 0.1
complex	$k_1 \; (\min^{-1})$ for conformational change		$k_{\rm 2C}~({\rm min}^{-1})$ for Fe ³⁺ release
Fe _C hTF/sTFR ^a	20.6 ± 1.2		7.2 ± 0.4
H349A $Fe_ChTF/sTFR^d$	9.2 ± 0.8		1.4 ± 0.5
$Fe_{C}hTF/sTFR \Delta 757-760$	48.9 ± 3.7		3.2 ± 0.1
complex	$k_1~({ m min}^{-1})$ for conformational change		$k_{\rm 2N}~({\rm min}^{-1})$ for Fe ³⁺ release
Fe _N hTF/sTFR ^a	22.0 ± 0.7		1.7 ± 0.6
Fe _N hTF/sTFR Δ 757–760	31.9 ± 0.5		3.4 ± 0.1

"Averages and 95% confidence intervals for kinetic runs performed on three to five different days. Each day, three kinetic traces were averaged before fitting." From ref 12. "From ref 15. "From ref 17.



Fe2hTF FechTF FenhTF Fe2hTF FechTF FenhTF Fe2hTF FechTF Fe

Figure 4. Urea gel analysis of hTFs in the presence of the sTFR and sTFR mutants. Samples were electrophoresed before (-) and after (+) incubation with iron removal buffer [100 mM MES (pH 5.6) containing 300 mM KCl and 4 mM EDTA] for 5 min.

subjected to iron removal buffer. Thus, the release of iron from this construct proceeds by way of a single pathway $(k_{1N} \rightarrow k_{2C})$, eliminating the need to include the other pathway $(k_{1C} \rightarrow k_{2N})$ in fitting the kinetic data for the Fe₂hTF/sTFR Δ 757–760 complex.

As mentioned, the release of iron from the Fe_chTF/sTFR complex (k_{2C}) is preceded by a conformational change (k_1). While the rate constant for the conformational change is markedly increased (48.9 min⁻¹) in the Fe_chTF/sTFR Δ 757–760 complex relative to the control Fe_chTF/sTFR complex (20.6 min⁻¹), the rate constant k_{2C} for iron release is significantly decreased [from 7.2 to 3.2 min⁻¹ (Table 3)]. Additionally, in the Fe_NhTF/sTFR Δ 757–760 complex, the rate constant for the initial conformational change (k_1) is also increased (from 22.0 to 31.9 min⁻¹), but the rate constant for iron release (k_{2N}) is doubled (3.4 min⁻¹) in comparison to that of the Fe_NhTF/sTFR control (1.7 min⁻¹) (Table 3). Thus, opposite effects of the Δ 757–760 mutation on iron release are seen for the two monoferric hTFs.

Kinetic Analysis of the Fe_chTF/sTFR Δ 757–760 Complex as a Function of pH. The pH sensitivity of the Fe_chTF/sTFR Δ 757–760 complex was examined by monitoring the iron release kinetics of the complex between pH 5.6 and 6.2. Both rate constants, k_1 and k_{2C} , of the Fe_chTF/sTFR complex are sensitive to pH: the rate constant reporting the conformational change (k_1) is decreased as the pH increases from 5.6 to 6.2 (Figure 5, inset), while iron release does not occur at pH \geq 6.0 (Figure 5). The pH profile of the Fe_chTF/ sTFR Δ 757–760 complex differs drastically in comparison to the control: the already rapid rate constant for conformational change increases slightly from pH 5.6 to 6.0, as does the rate constant for iron release (Figure 5).

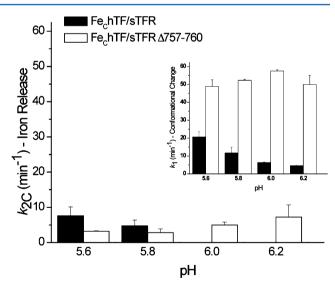


Figure 5. Effect of pH on rate constants of Fe_chTF/sTFR and Fe_chTF/sTFR Δ 757–760 complexes. Rate constants [k_1 for conformational change (inset) and k_{2C} for iron release] \pm 95% confidence intervals as a function of pH are shown for the Fe_chTF/sTFR control (black) and the Fe_chTF/sTFR Δ 757–760 complex (white).

DISCUSSION

The homodimeric TFR ectodomain is comprised of three distinct domains in each monomer. The protease-like domain (domain I, residues 121-188 and 384-606) contains two of the four Ca²⁺ binding residues [Glu465 and Glu468 (Figure 1B)] mutated in this work as well as one of the histidines

[His475 (Figure 1C)] comprising half of the histidine cluster that forms as a result of hTF binding. The apical domain (domain II, residues 189–383), in which a long loop (residues 275–338) containing His318 resides (Figure 1E), is not directly involved in binding hTF. When hTF binds to the ectodomain of the TFR, part of this loop markedly changes its position, moving into the proximity of the C-terminal residues of the other TFR monomer (Figure 1F).¹⁵ Specifically, Phe316' moves 8 Å, while His318' flips directly into the TFR–TFR′–C1 intersection (a movement of nearly 18 Å), bringing it within 5 Å of the C-terminus of the TFR.¹⁵ Lastly, the helical domain (domain III, residues 607–760), responsible for dimerization, contains the other histidine [His684 (Figure 1C)] in the interface histidine cluster and the four C-terminal amino acids of the TFR [Asp757, Asn758, Glu759, and Phe760 (Figure 1E)].

Glu465 and Glu468 in the protease-like domain of the sTFR and Asp307, Thr310, and Phe313 in the apical domain participate in the octahedral coordination of a metal ion (identified as Ca^{2+} in BHK-derived sTFR) (Figure 1B). The essential role of the bound metal in stabilizing the structure of the TFR is demonstrated by the poor production of the [E465A,E468A] sTFR mutant and the inability to isolate any nonaggregated sTFR, preventing further experimentation with this mutant. Although the exact role of the metal in vivo remains unclear, our results suggest that the coordination of this Ca^{2+} is important to the structural integrity of the sTFR.

The histidine cluster (His475 and His684 from each TFR monomer) formed when either HFE or hTF binds to the TFR (Figure 1C) has been suggested to be a pH-inducible motif that may trigger a conformational event in the TFR involved in the release of HFE or iron from hTF within the endosome.^{15,16} However, as previously noted,¹⁵ this histidine cluster, buried deep within the interface between the two TFR monomers, may be inaccessible to changes in pH. The relatively small differences in most of the kinetic parameters in the [H475A,H684A] sTFR complexes (Table 2) are inconsistent with the suggestion that this cluster is involved in the mechanism of pH-induced iron release, at least under the tested conditions (pH 5.6 with 300 mM KCl and 4 mM EDTA).

As mentioned, because of the inherent flexibility of the Cterminus, at neutral pH the final two amino acids of the TFR (Glu759 and Phe760) are not observed in the Fe_NhTF/sTFR crystal structure.¹⁵ Significantly, the C-terminus of one TFR monomer is located in the center of the TFR–TFR′–C1 intersection, directly between TFR′ residue His318 and hTF residue His349 shown to be a pH-inducible switch required for TFRstimulated release of iron from the C-lobe (Figure 1E).^{15,18} We propose that the proximity of these two pH-sensitive histidine residues to the C-terminus of the other TFR monomer provides an explanation for the many rather subtle kinetic effects discussed below.

A curious feature of the kinetic profiles of all hTF/sTFR complexes is the initial drop in the magnitude of the fluorescence signal as a result of the decrease in pH to 5.6. We have previously reported this phenomenon²¹ and, because it is observed only in sTFR-containing samples and sTFR alone, assigned it to a pH-induced conformational change in the sTFR. The significantly longer duration of this feature in the H318A sTFR mutant alone or complexed with Fe₂hTF is clearly shown in Figure 3 and is consistent with protonated His318 promoting and reporting this change. When hTF binds to the ectodomain of the TFR, His318 moves into the interface, close to TFR residues Trp641 and Trp740 of the TFR'

monomer [~3.6 and 4.1 Å, respectively (Figure 1 of the Supporting Information)]. Because protonated histidine residues quench tryptophan fluorescence by electron transfer,²⁵ it is reasonable to assign the pH-induced decrease in fluorescence of sTFR largely to His318 quenching of nearby Trp641 and/or Trp740 at low pH.

A new and interesting finding is that the binding of Fe₂hTF significantly prolongs the duration of the fluorescence quenching event in the sTFR (~3.4-fold), a phenomenon that is even more pronounced in the presence of the sTFR H318A mutation [~6.4-fold (Figure 3)]. In contrast, binding of Fe₂hTF does not increase the duration of the pH-inducible change in the truncated sTFR Δ 757–760 (Figure 3). On the basis of these results, we suggest that the C-terminus of the TFR interacts with His318' of the TFR' at neutral pH either through a hydrogen bond with Asn758 or through a weak hydrophobic interaction with Phe760, but only when ironcontaining hTF is bound. This interaction between the Cterminus and His318 is likely destabilized at low pH (5.6), thereby freeing the C-terminus to interact with hTF via residue His349. Further support for this suggestion is provided by the kinetics of the release of iron from the H318A sTFR-containing complexes (Table 1). Specifically, the conformational change that is normally found prior to the release of iron from the Clobe is not observed in the Fe_chTF/H318A sTFR complex. In a manner reminiscent of the H349W or H349Y FechTF mutant/sTFR complexes, which also lack the conformational change,¹⁸ the release of iron from the Fe_ChTF/H318A sTFR complex can occur without the pH-induced conformational change. Thus, elimination of the His318-C-terminus interaction as in the H318A sTFR mutant allows the C-terminus to interact with His349 of hTF even at neutral pH.

Whereas the conformational change preceding the release of iron from the Fe_ChTF/sTFR complex is not observed in the Fe_ChTF/H318A sTFR complex, it is accelerated more than 2-fold in the Fe_ChTF/sTFR Δ 757–760 complex [k_1 = 20.6 vs 48.9 min⁻¹ (Table 3)]. This finding supports the link between His318 of TFR' and the C-terminus of the other TFR monomer and provides further evidence that the two interact and have an effect on the conformational change.

In contrast to the Fe_ChTF/H318A sTFR complex, the conformational change is observed and little affected in the other monoferric Fe_NhTF/H318A sTFR complex (Table 1). These results imply that the conformational change preceding iron release in the two monoferric complexes has a different physical basis despite having nearly identical rate constants [20.6 min⁻¹ for the Fe_ChTF/sTFR complex vs 22.0 min⁻¹ for the Fe_NhTF/ sTFR complex (Table 1)].

Interestingly, because the C-terminus of the sTFR (residues 757–760) is only proposed to interact with the C1 subdomain of the C-lobe of hTF and residues of the protease-like domain of the other TFR monomer, it was expected that truncation of the C-terminus of the sTFR would not affect the release of iron from the N-lobe of Fe_NhTF. However, the rate constant for the initial conformational change (k_1) is increased by 45%, and the rate constant for iron release (k_{2N}) is doubled in the Fe_NhTF/ sTFR Δ 757–760 complex in comparison to that of the Fe_NhTF/sTFR control (Table 3). These results indicate communication between the two lobes of hTF within the hTF/TFR complex. Given that His318 interacts with the C-terminus of the other TFR monomer at pH 7.5, removal of the C-terminus eliminates this interaction, allowing His318 to interact with Trp641 and/or Trp740 even at neutral pH

(Figure 1 of the Supporting Information). Of possible significance, Trp641 is located on α III-3 in the helical domain of the TFR that interacts with both the N1 and C1 subdomains in the Fe_NhTF/sTFR crystal structure.¹⁴ Thus, we postulate that the absence of the interaction between His318 and the C-terminus in the Fe_NhTF/sTFR Δ 757–760 complex accelerates both the rate of conformational change and the rate of release of iron from the N-lobe in comparison to the Fe_NhTF/sTFR control via the interaction between His318 and Trp641 within α III-3.

It is well established that a pH-induced conformational change involving hTF residue His349 drives the release of iron from the C-lobe.^{14,17} A previous in silico model²⁶ of the hTF/TFR complex suggested that hTF C1 subdomain residue His349 interacts with hTF C2 subdomain residue Lys511 through a weak electrostatic interaction between the lone pair of electrons on ND1 of His349 and the cation of Lys511. However, because of the ~4.0-5.0 Å distance between His349 and Lys511 in the recent crystal structure of diferric hTF (PDB entry 3V83), this interaction is unlikely. Interestingly, mutation of Lys511 to alanine increases the rate of release of iron from the C-lobe, by an unknown mechanism.¹⁶ Closer examination of the crystal structure of diferric hTF reveals that Lys511 lies within 3.2 Å of and likely forms a salt bridge with hTF C1 subdomain residue Glu372, possibly explaining the observed effects of the K511A mutation.

Notably, the rate constant for the release of iron from the control Fe_chTF/sTFR complex titrates with an apparent pK_{a} of 5.8-5.9, consistent with titration of one or more histidine residues (specifically hTF residue His349).¹⁸ Intriguingly, the pH profile of the rate constants for the Fe_chTF/sTFR Δ 757– 760 complex (Figure 5) follows the same trend as the pH profile for the H349A Fe_chTF/sTFR complex.¹⁸ Moreover, the kinetics of the release of iron from the Fe₂hTF/sTFR Δ 757-760 complex (Table 3) are very similar to the published kinetic data for the H349A Fe₂hTF/sTFR complex:¹⁵ both exclusively follow the $k_{1N} \rightarrow k_{2C}$ pathway preceded by a rapid conformational change (Table 3). Additionally, the rate constant for the release of iron from the C-lobe, k_{2C} , is markedly reduced in the Fe₂hTF/sTFR Δ 757–760 complex. Hence, the inability to anchor the protonated His349 (as in the H349A hTF or the truncated sTFR) has a pronounced effect on iron release. These findings strongly support the suggestion that the C-terminus of the TFR stabilizes protonated His349 and promotes the release of iron from the C-lobe of hTF at pH 5.6 either through a cation $-\pi$ interaction with the C-terminal Phe760 of the sTFR or through formation of a salt bridge with Glu759 or Asp757. Individual mutation of the four C-terminal residues is required to establish the contribution of each to the TFR-facilitated mechanism of the release of iron from the C-lobe of hTF.

In summary, these data suggest that the release of iron from the hTF/TFR complex is controlled by a relay within the TFR–TFR'–C1 intersection (Figure 1E). The absence of the coordinated Ca^{2+} ion (Figures 1B and 2) or a glycosylation site in this region (Asn317)²³ has a negative effect on the structural stability of this region. At pH 7.5, His318' (TFR') interacts with the C-terminus of the other TFR monomer (Figure 1E). Upon exposure to the acidic environment within the endosome, the interaction between His318' and the C-terminus is severed by the protonation of His318', allowing His318' to interact with and quench nearby tryptophan residues 641 and/or 740 (Figure 3 and Figure 1 of the Supporting Information). A conformational change that promotes receptor-stimulated release of iron from the C-lobe of hTF (Figures 4 and 5) is then triggered by the interaction formed between protonated hTF residue His349 and the C-terminus of the TFR. Communication of these kinetic events to the N-lobe of hTF, possibly through helix α III-3 of the TFR to which both the N1 and C1 subdomains are bound, prompts the release of iron from the N-lobe. Collectively, these results help to establish a molecular basis for the pH-induced events that dictate efficient release of iron from each lobe within the endosome in a physiologically relevant time frame.

ASSOCIATED CONTENT

Supporting Information

Derivation of an equation for an A \rightarrow B model, including initial decay, the recombinant production levels of all sTFR mutants, oligonucleotide primers used to prepare the sTFR mutants, a representation of the Fe_NhTF/sTFR crystal structure showing the location of the TFR–TFR'–C1 intersection, including residues Trp641 and Trp740, and an example of the analysis of the initial quench in tryptophan fluorescence (fit to the A \rightarrow B model that also includes the initial decay term). This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*Department of Biochemistry, University of Vermont, Burlington, VT 05405. Telephone: (802) 656-0343. Fax: (802) 862-8229. E-mail: anne.mason@uvm.edu.

Funding

This work was supported by U.S. Public Service Grant R01 DK 21739 to A.B.M. and National Institute of General Medical Sciences Grant R37-GM-20194 to N.D.C. A.N.S. is funded by an AHA Predoctoral Fellowship (10PRE4200010).

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank Joe Klaus for technical assistance with the immunoblot and Dr. Brian E. Eckenroth for helpful discussions.

ABBREVIATIONS

hTF, human serum transferrin; TFR, transferrin receptor; apohTF, iron-free hTF; Fe₂hTF, recombinant N-terminal hexa-His-tagged nonglycosylated diferric hTF; sTFR, glycosylated N-terminal hexa-His-tagged soluble recombinant transferrin receptor (residues 121–760); Fe_NhTF, recombinant Nterminal hexa-His-tagged nonglycosylated monoferric hTF that binds iron only in the N-lobe (Y426F and Y517F mutations prevent iron binding in the C-lobe); Fe_ChTF, recombinant N-terminal hexa-His-tagged nonglycosylated monoferric hTF that binds iron only in the C-lobe (Y95F and Y188F mutations prevent iron binding in the N-lobe); BHK cells, baby hamster kidney cells; $t_{1/2}$, half-life; UG, Ultroser G; V_0 , void volume.

REFERENCES

(1) Hall, D. R., Hadden, J. M., Leonard, G. A., Bailey, S., Neu, M., Winn, M., and Lindley, P. F. (2002) The crystal and molecular structures of diferric porcine and rabbit serum transferrins at resolutions of 2.15 and 2.60 Å, respectively. *Acta Crystallogr. D58*, 70–80.

(2) Wally, J., Halbrooks, P. J., Vonrhein, C., Rould, M. A., Everse, S. J., Mason, A. B., and Buchanan, S. K. (2006) The crystal structure of iron-free human serum transferrin provides insight into inter-lobe communication and receptor binding. *J. Biol. Chem.* 281, 24934–24944.

(3) Fenton, H. J. H. (1876) On a new reaction of tartaric acid. *Chem. News* 33, 190.

(4) Fenton, H. J. H. (1893) The oxidation of tartaric acid in presence of iron. *Chem. Soc. Proc.* 9, 113.

(5) Aisen, P., Enns, C., and Wessling-Resnick, M. (2001) Chemistry and biology of eukaryotic iron metabolism. *Int. J. Biochem. Cell Biol.* 33, 940–959.

(6) Klausner, R. D., Van Renswoude, J., Ashwell, G., Kempf, C., Schechter, A. N., Dean, A., and Bridges, K. R. (1983) Receptormediated endocytosis of transferrin in K562 cells. *J. Biol. Chem.* 258, 4715–4724.

(7) Bali, P. K., Zak, O., and Aisen, P. (1991) A new role for the transferrin receptor in the release of iron from transferrin. *Biochemistry* 30, 324–328.

(8) Ohgami, R. S., Campagna, D. R., Greer, E. L., Antiochos, B., McDonald, A., Chen, J., Sharp, J. J., Fujiwara, Y., Barker, J. E., and Fleming, M. D. (2005) Identification of a ferrireductase required for efficient transferrin-dependent iron uptake in erythroid cells. *Nat. Genet.* 37, 1264–1269.

(9) Gunshin, H., Mackenzie, B., Berger, U. V., Gunshin, Y., Romero, M. F., Boron, W. F., Nussberger, S., Gollan, J. L., and Hediger, M. A. (1997) Cloning and characterization of a mammalian proton-coupled metal-ion transporter. *Nature* 388, 482–488.

(10) Dautry-Varsat, A., Ciechanover, A., and Lodish, H. F. (1983) pH and the recycling of transferrin during receptor-mediated endocytosis. *Proc. Natl. Acad. Sci. U.S.A.* 80, 2258–2262.

(11) Leverence, R., Mason, A. B., and Kaltashov, I. A. (2010) Noncanonical interactions between serum transferrin and transferrin receptor evaluated with electrospray ionization mass spectrometry. *Proc. Natl. Acad. Sci. U.S.A.* 107, 8123–8128.

(12) Byrne, S. L., Chasteen, N. D., Steere, A. N., and Mason, A. B. (2010) The unique kinetics of iron-release from transferrin: The role of receptor, lobe-lobe interactions and salt at endosomal pH. *J. Mol. Biol.* 396, 130–140.

(13) Lawrence, C. M., Ray, S., Babyonyshev, M., Galluser, R., Borhani, D. W., and Harrison, S. C. (1999) Crystal structure of the ectodomain of human transferrin receptor. *Science* 286, 779–782.

(14) Cheng, Y., Zak, O., Aisen, P., Harrison, S. C., and Walz, T. (2004) Structure of the human transferrin receptor-transferrin complex. *Cell* 116, 565–576.

(15) Eckenroth, B. E., Steere, A. N., Chasteen, N. D., Everse, S. J., and Mason, A. B. (2011) How the binding of human transferrin primes the transferrin receptor potentiating iron release at endosomal pH. *Proc. Natl. Acad. Sci. U.S.A.* 108, 13089–13094.

(16) Bennett, M. J., Lebron, J. A., and Bjorkman, P. J. (2000) Crystal structure of the hereditary haemochromatosis protein HFE complexed with transferrin receptor. *Nature* 403, 46–53.

(17) Steere, A. N., Miller, B. F., Roberts, S. E., Byrne, S. L., Chasteen, N. D., Smith, V. C., MacGillivray, R. T. A., and Mason, A. B. (2012) Ionic residues of human serum transferrin affect binding to the transferrin receptor and iron release. *Biochemistry* 51, 686–694.

(18) Steere, A. N., Byrne, S. L., Chasteen, N. D., Smith, V. C., MacGillivray, R. T., and Mason, A. B. (2010) Evidence that His349 acts as a pH-inducible switch to accelerate receptor-mediated iron release from the C-lobe of human transferrin. *J. Biol. Inorg. Chem.* 15, 1341–1352.

(19) Steere, A. N., Byrne, S. L., Chasteen, N. D., and Mason, A. B. (2012) Kinetics of iron release from transferrin bound to the transferrin receptor at endosomal pH. *Biochim. Biophys. Acta* 1820, 326–333.

(20) Bali, P. K., and Aisen, P. (1992) Receptor-induced switch in sitesite cooperativity during iron release by transferrin. *Biochemistry 31*, 3963–3967. (21) James, N. G., Byrne, S. L., and Mason, A. B. (2008) Incorporation of 5-hydroxytryptophan into transferrin and its receptor allows assignment of the pH induced changes in intrinsic fluorescence when iron is released. *Biochim. Biophys. Acta* 1794, 532–540.

(22) James, N. G., Byrne, S. L., Steere, A. N., Smith, V. C., MacGillivray, R. T., and Mason, A. B. (2009) Inequivalent contribution of the five tryptophan residues in the C-lobe of human serum transferrin to the fluorescence increase when iron is released. *Biochemistry* 48, 2858–2867.

(23) Byrne, S. L., Leverence, R., Klein, J. S., Giannetti, A. M., Smith, V. C., MacGillivray, R. T., Kaltashov, I. A., and Mason, A. B. (2006) Effect of glycosylation on the function of a soluble, recombinant form of the transferrin receptor. *Biochemistry* 45, 6663–6673.

(24) Mason, A. B., Halbrooks, P. J., Larouche, J. R., Briggs, S. K., Moffett, M. L., Ramsey, J. E., Connolly, S. A., Smith, V. C., and MacGillivray, R. T. (2004) Expression, purification, and characterization of authentic monoferric and apo-human serum transferrins. *Protein Expression Purif.* 36, 318–326.

(25) Callis, P. R., and Liu, T. (2004) Quantitative prediction of fluorescence in proteins. J. Phys. Chem. B 108, 4248-4259.

(26) Sakajiri, T., Yamamura, T., Kikuchi, T., and Yajima, H. (2009) Computational structure models of apo and diferric transferrintransferrin receptor complexes. *Protein J.* 28, 407–414.